

A radioactive isotope has a half-life of 7 years.

a) Use the formula  $A = A_0 C^p$ , with  $C = \frac{1}{2}$ , to determine how much of the isotope must initially be present to decay to 60 grams in 14 years.

A = 60

A = A0 CTP

A = A0

C = 
$$\frac{1}{3}$$
 $60 = A_0 \left(\frac{1}{4}\right)^{14/2}$ 
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b) Write an equivalent formula with C = 2 which would solve the problem in a).

c) In this particular example, explain how the solution to the problem in a) could be found without using any formula.

14475 = 2(halflife) -> intial amount reduces by 1/4 the by another 'ld ('/4) -> The final amount is '14 of the original so original 4x the final amount 4x60-2409

Complete Assignment Questions #8 - #15

## **Assignment**

- 1. A truck bought for \$35 000 depreciates at a rate of 12% per year.
  - a) If the value of the truck is an exponential function of time, state the base of the exponential function.  $1 0 \cdot 1 = 0 \cdot \%$
- b) Write an equation to represent the value, V, of the truck after t years.  $V = 35000 (0.88)^{-1} = 20989.33 \cdot \text{Val} \omega = 121000$ c) Determine, to the pearest hundred dollars, the value of the truck after t
  - c) Determine, to the nearest hundred dollars, the value of the truck after 4 years.  $\sqrt{-35000} (0.88)^{+}$
  - d) How many years, to the nearest tenth, would it take for the value of the truck to reduce to one quarter of its purchase price?

$$\frac{1}{4}35000 = 8150$$
  
 $8750 = 35000 (0.88)^{\dagger}$   
 $0.d5 = 0.88^{\dagger}$ 

$$t = \frac{\log 0.88}{\log 0.88}$$
  
 $t = \frac{\log 0.88}{\log 0.88}$   
 $t = 10.844...$ 



- 2. In 2008 the world population was approximately 6.7 billion and was increasing at an annurate of 1.3%.
  - a) If the function representing the population, in billions, is of the form  $y = ab^x$ , state values for a and b.

b) Write an equation to represent the world population, W billions, as a function of the number of years, n, since 2008. W=6.7 (1.013)

d) If the population continues to grow at this rate, determine the number of years, to the nearest year, for the population to double from its 2008 size.

6.7x2 = 13.4  

$$13.4 = 6.7(1.013)^n$$
  $\begin{cases} 1091.013 = 1091.013 \\ 1091.013 = 1091.013 \end{cases}$   $\begin{cases} 54.013 \\ 1091.013 \end{cases}$ 

e) Estimate the world population in 1950. State any assumptions you have made. How does your answer compare with the actual world population in 1950? Give a reason for any discrepancy.

- - a) Write an exponential function to represent the value of this robotic technology after t years. (maintains 752
  - b) How many years, to the nearest year, would it take for the value of the robotic technology, which initially cost \$575 000, to depreciate to \$25 000?

$$\frac{1}{33} = 0.75^{+}$$
 $\frac{1}{33} = 0.75^{+}$ 
 $\frac{1}{33} = 0.75^{+}$ 
 $\frac{1}{33} = 0.75^{+}$ 
 $\frac{1}{33} = 10.899...$ 

- 4. A town in southern British Columbia is growing at a rate of 3.5% per annum. If the town continues to grow at this rate, it is projected that the population will reach 20 000 in 5 years.
  - a) Determine, to the nearest ten people, the current population of the town.

$$P = \frac{d0000}{(1.035)^5} = 16839.4$$

-> current pop is 16840 people b) Assuming the same growth rate, determine how many years from now the popul will reach 30 000. Answer to the nearest year.

 $30\ 000 = 16840(1.035)^{n}$   $P \log_{1.035}\left(\frac{750}{421}\right) = n$   $30\ 000 = 1.035^{n}$  N = 16.78...

150 = 1.0357 5. A quantity of water contains 500 g of pollutants. Each time the water passes through a filter, 18% of the pollutants are removed. How many filters are needed to reduce the mass of pollutants to less than 150 g?

7 filters are

1090.82 0.3 = M

 $A = P(1+i)^n$  n = 1090.3 Thilters are 150 = 500 (0.8a) 1090.82 needed to reduct the mass to 1.20 = 0.3 =  $(0.8a)^n$  n = 6.066... reduct the mass to 1.20 =  $(0.8a)^n$  n = 6.066... rest than 150g. 155 than 150g.

- 6. An x-ray beam of intensity,  $I_0$ , in passing through absorbing material x millimeters thick, merges with an intensity, I; given by  $I = I_0 e^{-kx}$ . When the material is 9 millimetres thick, 50% of the intensity is lost.
  - a) Calculate the value of the constant k to the three decimal places.

$$0.5 = \frac{1}{6}e^{-9k}$$

$$10.5 = -9K$$

b) What percentage intensity, to one decimal place, remains if the

let I0=100 T=100e-0.077x 21.49, remains

HICKUSS)

- 7. A hot piece of metal loses heat according to the formula  $T = T_0 e^{-0.2t}$ , where T is the temperature difference between the metal and the surrounding air after t minutes and  $T_0$  is the initial temperature difference.
  - a) If the initial temperature of the metal was 330°C and of the air 30°C, find the temperature of the metal, to the nearest degree, after 5 minutes.

b) A different piece of hot metal cools to a temperature of 200°C after 8 minutes. What was the original temperature of the metal, to the nearest degree, if the air temperature

$$T = T_0 e^{-0.aT}$$

$$173 = T_0 e^{-0.a(8)}$$

$$T_0 = \frac{173}{e^{-1.6}}$$

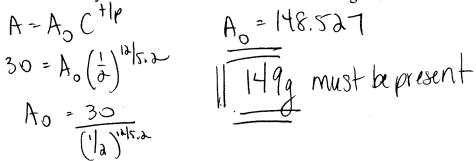
Temp diff =  $200^{\circ}$  -  $27^{\circ}$  =  $173^{\circ}$   $T = T_0 e^{-0.2}$   $T = T_0 e^{-0.2}$  original temp diff =  $857^{\circ}$  t = 8 t

8. How much of a radioactive substance must be present to decay to 30 grams in 12 years if the half-life of the substance is 5.2 years? Round the answer to the nearest gram.

A = 
$$A_0$$
 C +  $|P|$ 

30 =  $A_0$  ( $\frac{1}{a}$ )  $|a|_{5,a}$ 

A =  $\frac{30}{(|a|)^{4}}$   $|a|_{5,a}$ 



9. A radioactive isotope has a half-life of approximately 45 minutes. How long would it take for 480 mg of the isotope to decay to 15 mg?

$$A = A_0 C^{\dagger lp}$$
 $15 = 480(\frac{1}{a})^{1/45}$ 
 $15 = (\frac{1}{a})^{1/45}$ 
 $15 = (\frac{1}{a})^{1/45}$ 

$$|P| \Rightarrow \left(\frac{1}{a}\right)^{5} = \left(\frac{1}{a}\right)^{7/45}$$

$$5 = \frac{1}{45}$$

$$1 + \frac{1}{45}$$

$$1 + \frac{1}{45}$$

$$1 + \frac{1}{45}$$

$$2 + \frac{1}{45}$$

$$3 + \frac{1}{45}$$

$$4 + \frac{1}{45}$$

a) How long does it take for the cells to divide to produce 4096 cells?

A=4096

A=A<sub>0</sub>C+IP

A=1

Hogh=1(a)+IH

$$2^{12}=2^{114}$$
 $48=+$ 
 $48=+$ 

t=tb) At what time is the vial half full?

since doubling time is 4 mins, it will be half full 4 mins before it is full—

4 mins before 6 am is 5:56 am

c) At what time is the vial  $\frac{1}{16}$  full?

o =4min

The population of germs in a dirty bathtub doubles every 20 minutes. How long, to the nearest minute, would it take for the population to triple?

A= 3n
$$A_0 = n$$

$$C = 2$$

$$T = 4$$

$$A_0 = n$$

$$C = 3$$

$$T = 4$$

$$T = 40$$

$$T =$$

A radioactive isotope has a half-life of approximately 25 weeks. How much of a sample of 50 grams of the isotope would remain after 630 days? (Round the answer to the nearest hundredth of a gram.)

A=A

A=50g

C=
$$\frac{1}{3}$$
 $A=50(\frac{1}{a})^{630/115}$ 
 $A=50(\frac{1}{a})^{630/115}$ 

13. What is the half-life, to the neares month of a radioactive isotope if it takes 7 years for 560 grams to decay to 35 grams?

A= 35q
$$A = A_0 C + P$$

$$C = 1/2$$

$$A = 5600q$$

$$C = 1/2$$

$$A = 5600 \left(\frac{1}{a}\right)^{1/p}$$

$$A = 7$$

$$P = 7$$

$$S = 7$$



14. The tripling period, to the nearest tenth of an hour, of a bacterial culture which grows from 500 cells to 64 000 cells in 50 hours is \_\_\_\_\_.

(Record your answer in the numerical response box from left to right.)

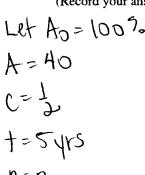
$$A = A_0 C^{+/p}$$
  
 $64000 = 500(3)$   
 $128 = 3^{50/p}$   
 $1093128 = 50$ 

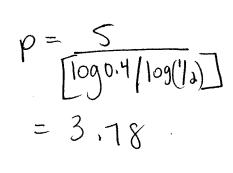
$$p = \frac{50}{\log_3 128}$$
=  $\frac{50}{(\log 128) \log_3}$ 

15. Radioactive material decays to 40% of its original mass in 5 years. The half-life of the radioactive material, to the nearest hundredth of a year, is \_\_\_\_\_.

(Record your answer in the numerical response box from left to right.)

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3		1	8	





## Answer Key

- 1. a) 0.88
- **b**)  $V = 35\ 000(0.88)^t$
- c) \$21 000
- **d)** 10.8

- **2.** a) a = 6.7, b = 1.013
- **b**)  $W = 6.7(1.013)^n$
- c) 8.3 billion
- **d**) 54 years
- e) 3.2 billion assuming a growth rate of 1.3% since 1950. The actual population was 2.55 billion, so the average growth rate since 1950 must have been greater than 1.3%.
- 3. a)  $V_t = V_0(0.75)^t$
- **b**) 11
- 4. a) 16 840
- **b**) 17 years
- 5. 7 filters

- **6. a)** 0.077
- **b**) 21.4%
- 7. a) 140°C
- **b**) 884°C
- 8. 149 grams

- 9 225 min
- 10.a) 48 min
- **b**) 5:56 a.m.
- c) 5:44 a.m.

- 11. 32 min
- 12. 4.12 g
- 13. 21 months

17.

- 18.
- 7